

# School of Basic and Applied Sciences

Course Code : BSCP3001

Course Name: QUANTUM MECHANICS

## Quantum Mechanics

### Covered Topics

- ❖ Wave velocity (or phase velocity) of de Broglie waves
- ❖ Group Velocity
- ❖ Wave packet
- ❖ Wave packet, phase velocity and group velocity
- ❖ References

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Name of the Faculty: Dr. ASHUTOSH KUMAR

Program Name: B.Sc. (Hon.) Physics

## Wave velocity (or phase velocity) of de Broglie waves

- The velocities of the individual waves which superpose to produce the wave packet representing the particle are different - the **wave packet as a whole** has a different velocity from the waves that comprise it

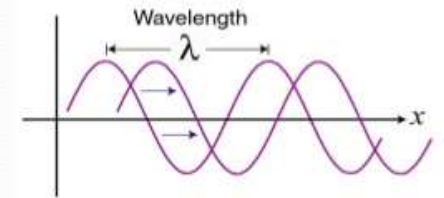
- **Phase velocity**: The rate at which the phase of the wave propagates in space

- **Group velocity**: The rate at which the envelope of the wave packet propagates

### The Phase Velocity

How fast is the wave traveling?

Velocity is a reference distance divided by a reference time.



The phase velocity is the wavelength / period:  $v = \lambda / \tau$

Since  $f = 1/\tau$ :

$$v = f \lambda$$

In terms of  $k$ ,  $k = 2\pi / \lambda$ , and the angular frequency,  $\omega = 2\pi / \tau$ , this is:

$$v = \omega / k$$

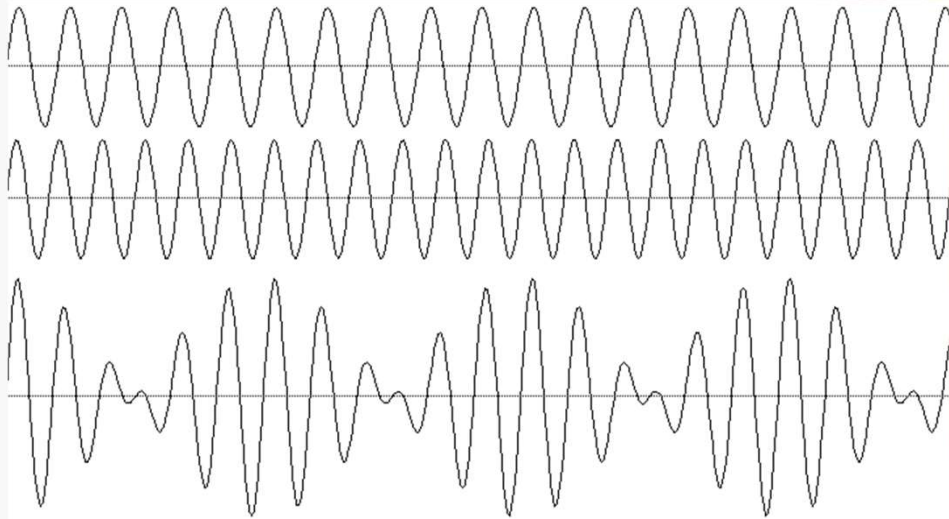
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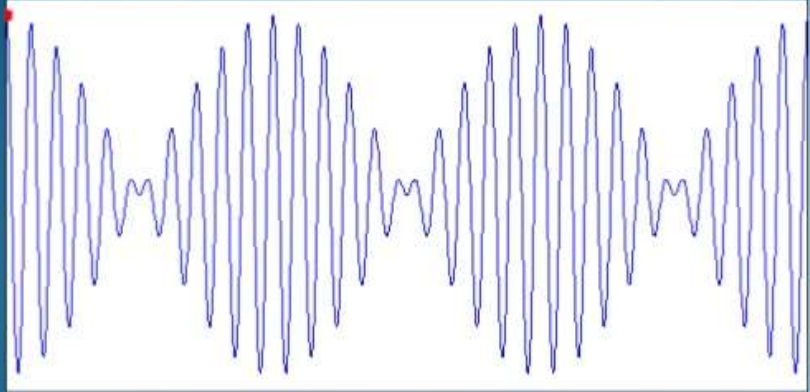
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## Group Velocity



**Adding up waves of different frequencies.....**

**The Group Velocity**



**This is the velocity at which the overall shape of the wave's amplitudes, or the wave 'envelope', propagates. (= *signal velocity*)**

**Here, phase velocity = group velocity (the medium is *non-dispersive*)**

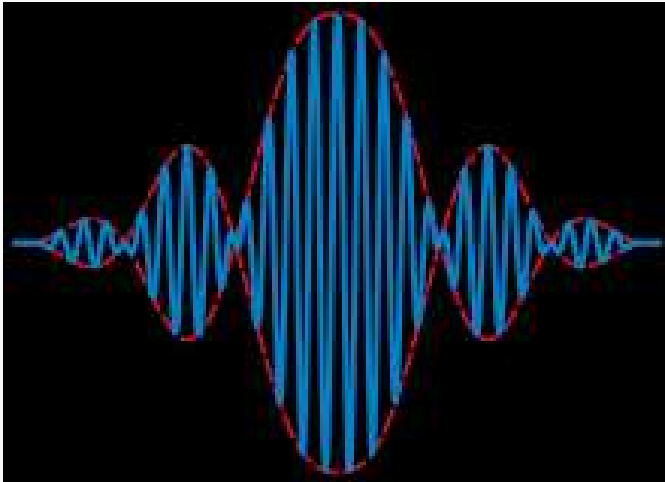
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## Wave packet

If several waves of different wavelengths (frequencies) and phases are superposed together, one would get a resultant which is a **localized wave packet**



A **wave packet** is a group of waves with slightly different wavelengths interfering with one another in a way that the amplitude of the group (envelope) is non-zero only in the neighbourhood of the particle

A wave packet is **localized** – a good representation for a particle!

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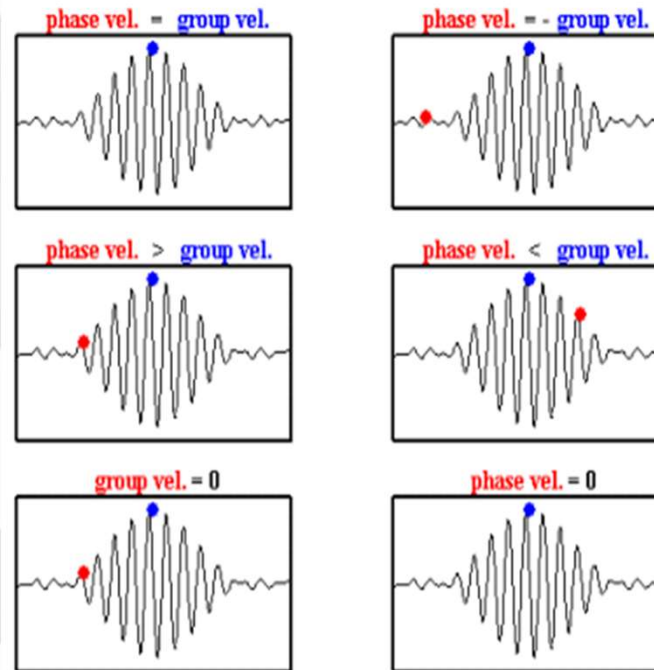
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## Wave packet, phase velocity and group velocity

- The velocities of the individual waves which superpose to produce the wave packet representing the particle are different - the **wave packet as a whole** has a different velocity from the waves that comprise it
- Phase velocity**: The rate at which the phase of the wave propagates in space
- Group velocity**: The rate at which the envelope of the wave packet propagates



### Phase velocity = Group Velocity

The entire waveform—the component waves *and* their envelope—moves as one. non-dispersive wave.

### Phase velocity = -Group Velocity

The envelope moves in the *opposite direction* of the component waves.

### Phase velocity > Group Velocity

The component waves move more quickly than the envelope.

### Phase velocity < Group Velocity

The component waves move more slowly than the envelope.

### Group Velocity = 0

The envelope is stationary while the component waves move through it.

### Phase velocity = 0

Now only the envelope moves over stationary component waves.

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## Wave packet, phase velocity and group velocity

### Phase velocity

$$v_p = \frac{\omega}{k} = \frac{E/\eta}{p/\eta} = \frac{E}{p}$$

$$v_p = \frac{E}{p} = \frac{\gamma mc^2}{\gamma mv} = \frac{c^2}{v} = \frac{c}{\beta}$$

$$v_p = \frac{c^2}{v}$$

Here  $C$  is the velocity of light and  $V$  is the velocity of the particle

$$v_g = v$$

$$v_g = v_p - \lambda \frac{dv_p}{d\lambda}$$

### Group velocity

$$v_g = \frac{\partial \omega}{\partial k} = \frac{\partial(E/\eta)}{\partial(p/\eta)} = \frac{\partial E}{\partial p}$$

$$v_g = \frac{\partial E}{\partial p} = \frac{\partial}{\partial p} \left( \frac{1}{2} \frac{p^2}{2m} \right)$$

$$= \frac{p}{m}$$

$$= v$$

$$v_g = \frac{\partial E}{\partial p} = \frac{\partial}{\partial p} \left( \sqrt{p^2 c^2 + m^2 c^4} \right)$$

$$= \frac{pc^2}{\sqrt{p^2 c^2 + m^2 c^4}}$$

$$= \frac{p}{m \sqrt{(p/(mc))^2 + 1}}$$

$$= \frac{m\gamma}{m\gamma v}$$

$$= v$$

21

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5. Modern Thermodynamics with Statistical Mechanics, Carl S. Helrich, Springer

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