

Curse Code : BSCP2051 Course Name: Laser Physics

CARBON DIOXIDE LASER (CO2 Laser)

CO₂ laser was one of the earliest <u>gas lasers</u> to be developed (invented by <u>C K N</u> <u>Patel</u> of <u>Bell Labs</u> in 1964), and is still one of the most useful.

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- Carbon Dioxide laser is a four-level molecular laser (operates on a set of vibrational-rotational transitions).
- Active Medium: Mixture of CO2, N2, He
- Emit in the Infra Red (IR) Spectrum ($\lambda = 9.6 \mu m \text{ or } 10.6 \mu m$)
- This laser is powerful enough to cut many substances and also destroy many others.
- As a continuous wave (CW) laser, this laser beam is the most powerful in production.
- Very simple to operate, and the gasses are non-toxic
- Pumping Method: Electrical excitation, Collisional transfer.

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Molecular Gas laser

In a molecular gas laser, laser action is achieved by transitions between vibrational and rotational levels of molecules. Its construction is simple and the output of this laser is continuous.

In CO2 molecular gas laser, transition takes place between the vibrational states of Carbon dioxide molecules.

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Energy states of CO2 molecules:

A carbon dioxide molecule has a carbon atom at the center with two oxygen atoms attached, one at both sides. Such a molecule exhibits three independent modes of vibrations. They are

- a) Symmetric stretching mode.
- b) Bending mode
- c) Asymmetric stretching mode.

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Construction:

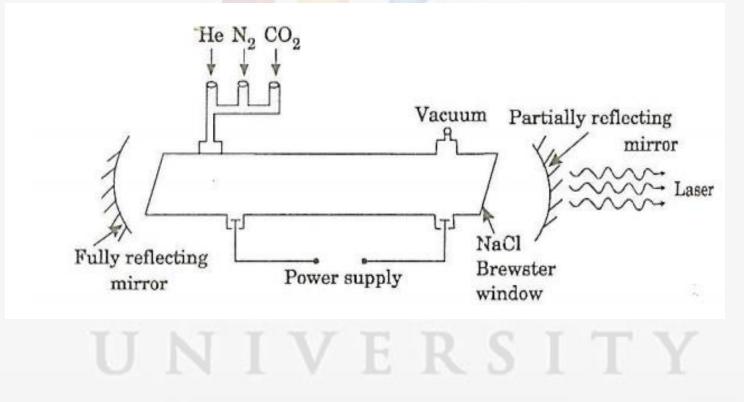
It consists of a quartz tube 5 m long and 2.5 cm in the diameter. This discharge tube is filled with gaseous mixture of CO2 (active medium), helium and nitrogen with suitable partial pressures.

The terminals of the discharge tubes are connected to a D.C power supply. The ends of the discharge tube are fitted with NaCl Brewster windows so that the laser light generated will be polarized. Two concave mirrors one fully reflecting and the other partially form an optical resonator.

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Construction:

The active medium is a gas mixture of CO₂, N₂ and He. The laser transition takes place between the vibrational states of CO₂ molecules



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Working of CO₂ Laser

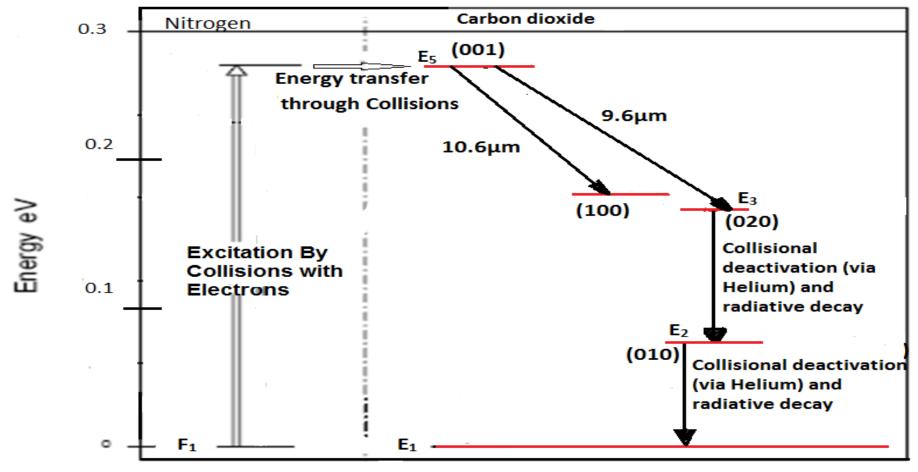
When electricity is run through the gas mixture, the particles of nitrogen become excited, meaning that they gain more energy.

Nitrogen is used because it can hold this excited state for long periods of time without discharging the energy in the form of photons, or light. (Nitrogen plays a similar role as that of He in He-Ne laser).

The excited vibrations of the nitrogen then cause the carbon dioxide to become excited to (E_5) level.

At this point, population inversion has been achieved between (E5) and (E4) giving stimulated emission (10.6μm) and (9.6μm) between (E5) and (E3) levels

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Relevant Energy levels of nitrogen and carbon dioxide

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Contd.....Working of CO2 Laser

- Helium is used to increase the thermal conductivity of the walls of the tube, efficiency by decreasing the population of (E3), (E2) levels and indirectly depleting the linked (E4) level.
- The light produced is so powerful compared to normal light because the tube of gases in a laser beam is surrounded by mirrors, which serve to reflect at least part of the light traveling through the tube. This reflection of light causes the light waves being produced by the nitrogen to reinforce themselves. This means that the light is amplifying as it travels through the gas tube, only coming out after reaching a certain intensity, making it extremely powerful.

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- Because of the high output power (combined with reasonable cost for the laser), CO₂ lasers are frequently used in industrial applications for <u>cutting</u>, <u>welding</u> and <u>hole</u> <u>drilling</u>
- In medical field Co₂ laser are used to destory infected tissue in a wound.
- Because the <u>atmosphere</u> is quite transparent to infrared light, CO₂ lasers are also used for military <u>range finding</u>

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• Easily shows the molecular properties of CO₂

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Reference Books:

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2. K. Thyagarajan, A. K. Ghatak, Lasers: Theory and Applications. New Delhi: Macmillan India Ltd (2011)

3. L. Allen, Essentials of Lasers. Oxford: Pergamon Press (2017)