Course Code : BSCP2005

Course Name: Elements of Modern Physics

UNIT 1: W&VE-P&RTICLE DU&LITY

The Uncertainty Principle: Photon Picture

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Course Name: Elements of Modern Physics

THE UNCERTAINTY PRINCIPLE

• One of the fundamental consequences of quantum mechanics is that it is IMPOSSIBLE to SIMULTANEOUSLY determine the POSITION and MOMENTUM of a particle with COMPLETE PRECISION

• Can be illustrated by a couple of "thought experiments", for example the "photon picture" of single slit diffraction and the "Heisenberg Microscope"

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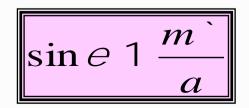
Course Code : BSCP2005 Course Name: Elements of Modern Physics SINGLE SLIT DIFFRACTION



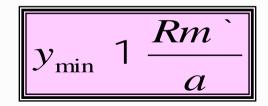
"geometrical" picture breaks down when slit wid with wavelength

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Course Code : BSCP2005 Course Name: Elements of Modern Physics POSITION OF DARK FRINGES IN SINGLE-SLIT DIFFRACTION

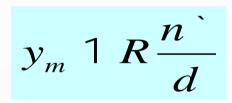


If, like the 2-slit treatment we assume small angles, sine $\approx \tan e = y_{\min}/R$, then



Positions of intensity <u>MINIMA</u> of diffraction pattern on screen, measured from central position.

Very similar to expression derived for 2-slit experiment:



But remember, in this case \boldsymbol{y}_m are positions of \underline{MAXIMA} In interference pattern

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WIDTH OF CENTRAL MAXIMUM

•We can define the width of the central maximum to be the distance between the m = +1 minimum and the m=-1 minimum:



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SINGLE SLIT DIFFRACTION: PHOTON PICTURE

Since θ is small:

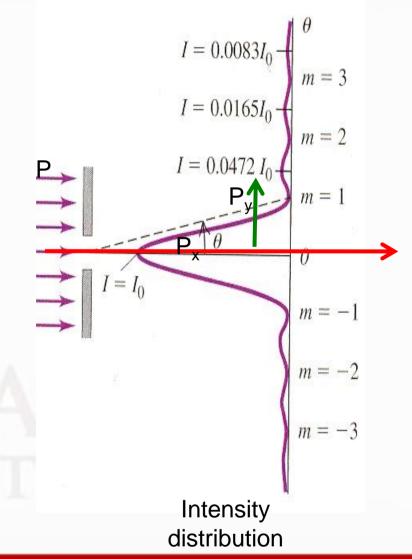
$$\sin e 1 - \frac{1}{a} \quad e 1 - \frac{1}{a}$$

Photons directed towards outer part of central maximum have momentum

$$\overline{p} \ 1 \ \overline{p}_x \ \check{Z} \ \overline{p}_y$$

$$p_y \ 1 \ e_{p_x} \ 1 \ p_x \frac{\dot{A}}{a} \ 1 \ p_x \frac{\dot{A}}{p_x a} \ 1 \frac{h}{a}$$

ie, localizing photons in the y-direction to a slit of width a leads to a spread of y-momenta of <u>at least</u> h/a.



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Course Code : BSCP2005

Course Name: Elements of Modern Physics

- So, the more we seek to localize a photon (ie define its position) by shrinking the slit width, *a*, the more spread (uncertainty) we induce in its momentum:
- In this case, we have $8p_y 8y r$ h

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Course Code : BSCP2005

Course Name: Elements of Modern Physics

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HEISENBERG MICROSCOPE

Suppose we have a particle, whose momentum is, initially, precisely known. For convenience assume initial p = 0.

From wave optics (Rayleigh Criterion)

$$\sin e - \frac{1}{D}$$

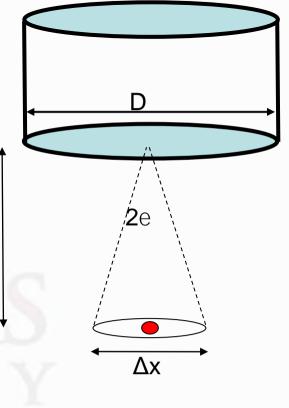
From our diagram:

agram:

$$\sin e^{\#} \frac{8x}{2} \oplus 1 \frac{8x}{2y}$$

$$8x^{\#} \frac{2y}{D}$$

"microscope"



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CP2005 Course Name: Elements of Modern Physics HEISENBERG MICROSCOPE

 $8x + \frac{2y}{D}$

Since this is a "thought experiment" we are free from any practical constraints, and we can locate the particle as precisely as we like by using radiation of shorter and shorter wavelengths. D // /2e

Δх

"microscope"

But what are the consequences of this?

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Program Name: B.Sc. Physics

V

Course Code : BSCP2005

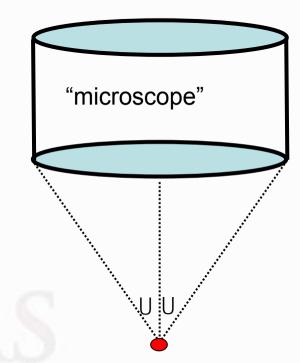
CP2005 Course Name: Elements of Modern Physics HEISENBERG MICROSCOPE

In order to see the particle, a photon must scatter off it and enter the microscope.

Thus process MUST involve some transfer of momentum to the particle......

BUT there is an intrinsic uncertainty in the X-component of the momentum of the scattered photon, since we only know that the photon enters the microscope somewhere within a cone of half angle U:

$$\Delta p = 2psinU$$



By conservation of momentum, there must be the same uncertainty in the momentum of the observed particle......

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Course Code : BSCP2005

Course Name: Elements of Modern Physics

HEISENBERG MICROSCOPE: SUMMARY

Uncertainty in position of particle:

$$8x + \frac{2y}{D}$$

Can reduce as much as we like by making mall

Uncertainty in momentum of particle: $8p " 2p_{photon} \sin U " \frac{2h}{2y} \frac{D}{2y}$

So, if we attempt to reduce uncertainty in position by decreasing λ , we INCREASE the uncertainty in the momentum of the particle!!!!!!

Product of the uncertainties in position and momentum given by:

$$8x8p " \frac{2y}{D} \frac{Dh}{y} 1 2h$$

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Course Code : BSCP2005

Course Name: Elements of Modern Physics

THE UNCERTAINTY PRINCIPLE

Our microscope though experiments give us a rough estimate for the uncertainties in position and momentum:

 $8x8p \sim h$

"Formal" statement of the Heisenberg u

$$8x8p \ddagger \frac{\hbar}{2}$$

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Course Code : BSCP2005

Course Name: Elements of Modern Physics

REFERENCES

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