School of Medical & Allied Sciences

Course Code: BPHT6001 Course Name: Medicinal Chemistry-III

SULPHONAMIDES

GALGOTIAS UNIVERSITY

Description

- One of the oldest antibacterial agents used to combat infection
- Used for coccal infection in 1935
- They are bacteriostatic because it inhibits bacterial synthesis of folic acid
- Clinical usefulness has decreased because of the effectiveness of other antibiotics and penicillin

Presence of free amino group



Prontosil red Prodrug

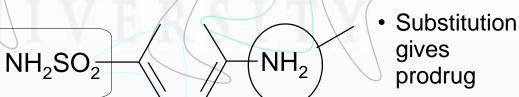
In vivo Active

$$N=N$$
 H_2N
 H_2N
 $Metablic cleavage$
 SO_2NH_2

$$\begin{array}{c|c} & & & & NH_2 \\ \hline & & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & & \\ & \\ & & \\ &$$

Sulphonamide

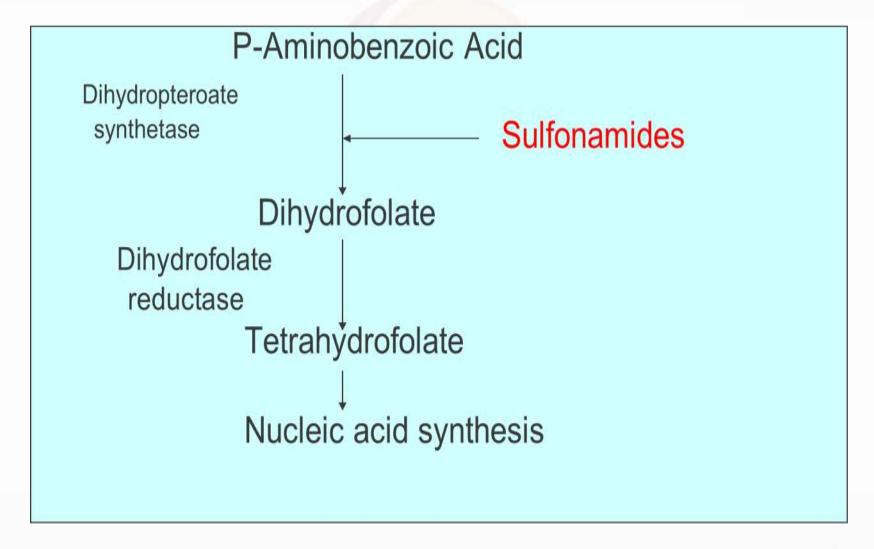
- ¬ Chemical modification of the sulphonamide structure has given rise to several important group of drugs.
 - ¬ Gloucoma − Acetazolamide
 - → Diuretic Thiazides
 - ¬ Anti-mycobacterial Sulphones
 - ¬ Oral hypoglycemic − Sulphonyl ureas
- 1) Sodium salt-- water soluble
 2) Substitution on these group gives different molecules having different pharmacokinetic properties



Mechanism of action

- Competitive inhibitor to dihydropteroate synthase enzyme due to resemblance with para-amino benzoic acid.
- Sulfonamides therefore are reversible inhibitors of folic acid synthesis and bacteriostatic not bactericidal.
- Inhibit bacterial growth without affecting normal cells

MECHANISM OF ACTION



Antibacterial activity

- · Gram-positive and gram negative.
- · Nocardia, chlamydia trachomatis, some protozoa.

Classification

A. Sulphonamides employed for treatment of systemic infection. Depending upon duration, they can be further subdivided into

a) Short to intermediate acting sulphonamides.

$$H_2N$$
 SO_2N
 O
 N
 O
 N

Sulphamethoxazole

$$H_2N$$
 SO_2N N N

Sulphadiazine

B. Long acting sulphonamides

C. Extra long acting sulphonamides

2. Poorly absorbed sulphonamides

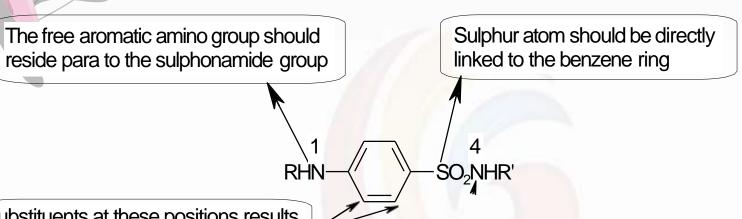
Mafenide acetate

3. Topically used sulphonamides

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STRUCTURE ACTIVITY RELATIONSHIP

- General
 - 1. Sulphonamide skeleton is the minimum structural requirement for antibacterial activity.
 - 2. The active form of sulphonamide is the ionized form. Maximum activity is observed bretween the pk_a value 6.6-7.4.
- 3. Sulphonamides competes for binding site on plasma albumin with causes increased action of drugs like



Substituents at these positions results in devoid of antibacterial activity

Substitution at this position activity varies with the nature of substituents.

- 1) Electron donating substituents to SO2 leads to increase in antibacterial activity.
- 2) Heterocyclic substituents leads to highly potent derivatives.
- 3) ubstitution of free Sulphonic acid (-SO3H) group for sulphonamide function destroys activity.
- 4) Replacement by a sulfinic acid group (-SO2H) an acetylation of N1 positio retains activity.

Therapeutic uses

- Urinary tract infections
- Upper respiratory tract infections
- Nocardiosis
- · Sulfasalazine in IBD.
- · Sulfacetamide in bacterial conjunctivitis & trachoma
- Silver sulfadiazine for prevention of infection of burn wounds.

Trimethoprim - Sulfamethoxazole combination (Co-trimoxazole)

Sulphamethoxazole

Trimethoprim

Mechanism of action:

- Sequential blocking of purine synthesis (synergism).
- Trimethoprim inhibits dihydrofolate reductase enzyme so inhibits tetrahydrofolic acid synthesis
- · The combination is bactericidal

Clinical uses

- Acute or Complicated or recurrent urinary tract infections especially in females
- Upper respiratory tract infections
- · Pneumocystis jiroveci pneumonia
- Toxoplasmosis
- Shigellosis
- Nocardiosis

Clinical uses continues.....

- Typhoid fever
- · Salmonella infections
- Prostatitis
- · Community -acquried bacterial pneumonia

Reference

- > William O. Foye., Textbook of Medicinal Chemistry, Pg. no: 1089 -1106
- Sriram., Medicinal Chemistry, Pg. no: 295-309.
- ➤ Kadam., Textbook of Medicinal Chemistry, Pg. no: 68-82.
- > Ilango., Principles of Medicinal chemistry(vol.1), Pg. no: 121-143.
- ≽Good man And GilMan's; The Pharmacology Basis Of Therapeutics Tenth Edition, pg. no 1189-1225.
- ➤JH Block & JM Beale., Wilson & Giswold's Textbook of Organic Medicinal Chemistry & pharmaceutical chemistry 12th Edition, 2011, pg. No. 260-294.